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Compounds with the α -U₃O₈-type structure.* By WILLIAM T. HOLSER,† *Battelle Memorial Institute, Columbus, Ohio, U.S.A.*

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Zachariassen (1948) described the structure of α -UO₃ (orange, hexagonal, Type I of Katz & Rabinowitz, 1951, p. 278) and mentioned that α -U₃O₈ had a similar structure, with a pseudo-hexagonal orthorhombic unit cell. Hoekstra, Siegel, Fuchs & Katz (1955) found that α -U₃O₈ was truly hexagonal only above 400 C. On the other hand Milne (1951) decided from Weissenberg data that his 'crystal' of U₃O₈ was a composite of two truly hexagonal forms with the same c_0 (4.15 Å) and different a_0 (3.94 Å and 3.87 Å), the change in a_0 being due to different oxygen contents. Brooker & Nuffield (1952) likewise interpreted most of their data as two hexagonal cells.

According to Elson, Fried, Sellers & Zachariassen (1950), γ -Nb₂O₅ (*T*-form of Brauer, 1941), β -Ta₂O₅ (low-temperature form of Magneli & Lagergren, 1952) and one form of Pa₂O₅ are orthorhombic and isostructural with U₂O₅ or UO_{2.6}, which is also homotypic (Megaw, 1946) with α -U₃O₈ and α -UO₃ (Baenziger, 1948; Hoekstra *et al.*, 1955). X-ray powder diffraction patterns, made in this laboratory, of γ -Nb₂O₅, β -Ta₂O₅ and α -U₃O₈ are all closely similar, in confirmation of published data. However, extension of Milne's interpretation of two hexagonal cells to the cases of Nb₂O₅ and Ta₂O₅ is inconsistent with the general chemistry of these elements, and with detailed knowledge of the Nb-O (Brauer, 1941) and Ta-O (Magneli & Lagergren, 1952) systems. Furthermore, detailed measurement and indexing of the powder diffraction patterns confirm the presence of an orthorhombic pseudo-hexagonal cell rather than two hexagonal cells (Table 1). Therefore, the orthorhombic type exists for all three compounds.

It should be noted that X-ray diffraction powder patterns of γ -Nb₂O₅ and β -Ta₂O₅ consistently show a few

weak lines at about 3.35, 2.55, 2.10 and 1.5 Å. These cannot be indexed on the described hexagonal unit cells, the orthorhombic unit cell (Table 1), or the latter with the a_0 , b_0 or c_0 parameter doubled. They apparently refer to a larger multiple cell, perhaps the very large cell found by Baenziger (1948) for UO_{2.6}. Single-crystal work will be required to identify these lines.

Some of the X-ray powder diffraction patterns referred to above were kindly furnished by D. A. Vaughan and J. R. Bridge of Battelle Memorial Institute.

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Table 1. X-ray powder diffraction data for compounds with the α -UO₃-type structure

Compound	a_0 (Å)	b_0 (Å)	c_0 (Å)	Non-hexagonal lines	References
U ₃ O ₈ *	6.70	3.97	4.14	(421) (600)	Gronwald (1948)
	6.71	3.96	4.15	(421) (600) (212) (210) (211) (213) (302?) (320?)	
γ -Nb ₂ O ₅ †	6.21	3.67	3.95	(211) (212)	Brauer (1941)
	6.19	3.65	3.94	(211)	
β -Ta ₂ O ₅ †	6.18	3.65	3.88	(210) (320?)	Vermilyea (1953)
	6.20	3.67	3.90	(210) (211) (212) (120) (302) (320?) (421?)	

* Milne's (1951) powder pattern (not taken on the same material for which he obtained Weissenberg data) does not even resolve some lines such as (110) and (111) which should be split in either interpretation, so it will not be considered further here. The General Electric pattern (ASTM card 2-0275) is nearly identical with Gronwald's data.

† The patterns by Peterson, Fassell & Wadsworth (1954, p. 1041) of Ta₂O₅ formed at temperatures between 500 and 1000 C. are poorly resolved, but two of them show the (211) line. Their pattern of 'ignited' Ta₂O₅ is the α (high-temperature) form. Hahn's (1951) patterns are also poorly resolved. There is a misprint apparent in his original publication, which is retained with added misprints on ASTM card 5-0258 and in Vermilyea (1953, p. 293). The pattern for Ta₂O₅ reported by Hanawalt *et al.* (1938) is a perovskite-type structure, apparently from soda contamination, as noted by Brauer for Nb₂O₅ and also experienced in this laboratory with Ta₂O₅.